

UNIVERSIDAI

COMPLUTENSE

MADRIE

# Analysis of soils in the Antarctic Region by calibration free-laser induced breakdown spectroscopy



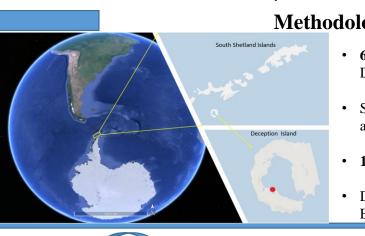
César Marina-Montes<sup>1</sup>, Roberto J. Lasheras<sup>1</sup>, Andrés Cruz-Conesa<sup>1</sup>, Abrahan Velásquez<sup>1</sup>, Jesús Anzano<sup>1</sup>, L.Vicente Pérez-Arribas<sup>2</sup>, Jorge Cáceres<sup>2</sup>, and Vincenzo Palleschi<sup>3</sup>

<sup>1</sup>Laser lab, Chemistry & Environment Group, Department of Analytical Chemistry, Faculty of Sciences, University of Zaragoza, Pedro Cerbuna 12, 50009-Zaragoza, Spain <sup>2</sup>Laser Chemistry Research Group, Department of Analytical Chemistry, Faculty of Chemistry, Complutense University of Madrid. Plaza de Ciencias 1, 28040-Madrid, Spain <sup>3</sup>Laboratory of Applied and Laser Spectroscopy, ICCOM/CNR, Via G. Moruzzi, 1, 56124 Pisa, Italy



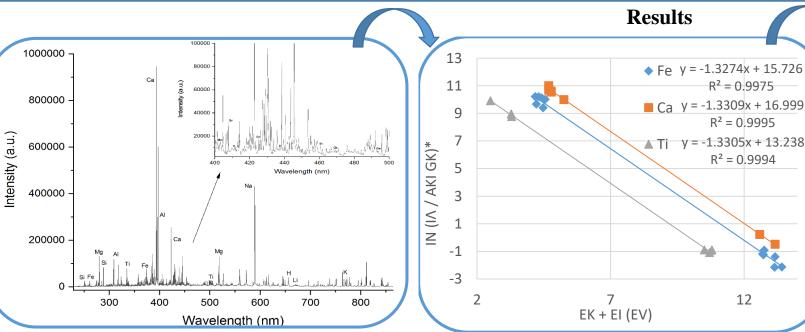
## Introduction

The study of geological samples, such as soils, is one of Laser-induced breakdown spectroscopy (LIBS) many applications [1]. Since sometimes the matrix is quite complex, it is complicated to find or manufacture standards for these type of samples. Thus, a good alternative is implementing Calibration Free (CF) [2]. When using CF, the physical parameters of the plasma created by the interaction of the laser with the sample, are studied and related with the elements and species that compose it. This methodology has been followed in this work to perform a multielemental quantitative analysis of soil samples from Antarctica, particularly at Deception Island (Gabriel de Castilla Spanish Research Base, South Shetland Islands).



## Methodology

- 6 soil samples from different parts of Deception Island were taken and analysed.
- Soil samples were pressed as **pellets** for LIBS analysis.
- 120 measurements per sample
- Delay Time = 1900 ns, Gate Width = 3650 ns, Energy = 45 mJ



No need of calibration standards. Estimation of Te and Ne to determine composition. Ne was obtained from the hydrogen line.

samples: Te =  $9024 \pm 54$  K Ne =  $8.3 \pm 0.1 \times 10^{16}$  cm<sup>-3</sup>

Oxides (%)	Sample 1	Sample 2	Sample 3	Sample 4	Sample 5	Sample 6
SiO <sub>2</sub>	$46.3\pm5$	$52.7\pm5.1$	$54.9 \pm 1.2$	$50.4 \pm 1.9$	$51.3\pm2.8$	$49.3\pm2.7$
Al <sub>2</sub> O <sub>3</sub>	$19.1 \pm 1.7$	$18.2\pm1.8$	$17.2\pm0.8$	$19.4\pm0.6$	$18.9\pm2$	$20.9\pm1.4$
Fe <sub>2</sub> O <sub>3</sub>	$11.8\pm1.4$	$9.9 \pm 1.4$	$9\pm0.2$	$9.7\pm0.3$	$9.8\pm0.7$	$10.9\pm0.6$
CaO	$9.03 \pm 1.28$	$7.7 \pm 1.2$	$7.3\pm0.5$	$8.1 \pm 1$	$8.2\pm0.8$	$7\pm0.9$
MgO	$5.8\pm1.9$	$5\pm1.2$	$4.5\pm0.5$	$4.9\pm0.2$	$4.1\pm0.3$	$4.3\pm0.6$
Na <sub>2</sub> O	$3.8\pm0.6$	$3\pm0.4$	$3.5\pm0.1$	$3.4\pm0.4$	$4\pm0.3$	$3.2\pm0.4$
TiO <sub>2</sub>	$3.3 \pm 0.3$	$2.6\pm0.4$	$3.1\pm0.2$	$3.4\pm0.4$	$3.1\pm0.3$	$3.6\pm0.4$
K <sub>2</sub> O	$1\pm0.2$	$0.9\pm0.4$	$0.7\pm0.1$	$0.7\pm0.1$	$0.7\pm0.1$	$0.8\pm0.1$

**Te** was estimated using the **Saha-Boltzman** plot. Mean of the 6 **LTE** verification using Mc Whirther Criterium:  $N_e$  (cm<sup>-3</sup>)  $\geq 1.6 \times 10^{12} T_e^{1/2} (\Delta E)^3$ 

## Conclusions

- The samples were mainly composed of the following oxides: SiO<sub>2</sub>, Al<sub>2</sub>O<sub>3</sub>, Fe<sub>2</sub>O<sub>3</sub>, CaO, MgO, Na<sub>2</sub>O, TiO and K<sub>2</sub>O.
- The results were compared with those provided by ICP-OES analysis obtaining close values for most oxides. For minority elements, the CF-LIBS and the ICP-OES results were within the same order of magnitude in all cases but the chrome case.
- These results show that CF-LIBS can be very useful in the characterization of complex samples, such as soils.

### References

[1] Rai, A.K., Pati, J.K., Rai, A.K., 2020. Chapter 16 - LIBS study of geological samples coupled with chemometric methods, in: Singh, J.P., Thakur, S.N. (Eds.), Laser-Induced Breakdown Spectroscopy (Second Edition). Elsevier, Amsterdam, pp. 369-384.

[2] Tognoni, E., Cristoforetti, G., Legnaioli, S., Palleschi, V., 2010. Calibration-Free Laser-Induced Breakdown Spectroscopy: State of the art. Spectrochimica Acta Part B: Atomic Spectroscopy 65, 1-14.

### Acknowledgment

This project forms part of the Ministry of Science, Innovation and Universities (Spain) proposal CTM2017-82929R in collaboration with the Government of Aragon proposal E23\_17D. Financial support from University of Zaragoza & Complutense University of Madrid is acknowledged. The authors thank the military staff at the Gabriel de Castilla Spanish research base for help with the installation of equipment and sample collection.

